淡江大學 107 學年度日間部寒假轉學生招生考試試題

系別: 化學系三年級

科目:有機化學

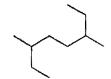


考試日期:1月13日(星期日)第2節

本試題共 2 大題, 3

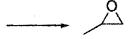
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- I. 選擇題 (20 pts, 2 pts each)
- 1. What is the parent chain for the following compound?

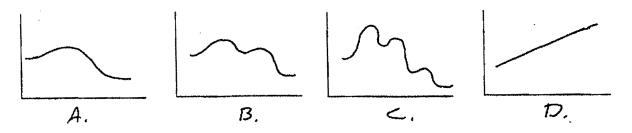


- A. octane
- B. hexane
- C. heptane
- D. decane
- 2. Why is entropy negative for ring closures?
- A. Closing a ring results in fewer molecules.
- B. Closing a ring results in more molecules.
- C. Closing a ring releases energy.
- D. Closing a ring restricts the rotation around individual carbon-carbon bonds.
- 3. Which of the following would you expect to have the most negative ΔS ?

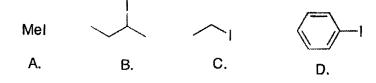
B. OH



4. Which of the following is an energy diagram for a two-step reaction?



5. Rank the following substrates from most to least reactive in an S_N2 reaction.



- A. A>B>C>D
- R D>C>R>A
- C. A>C>B>D
- D. D>C>B>A

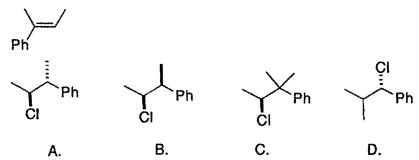
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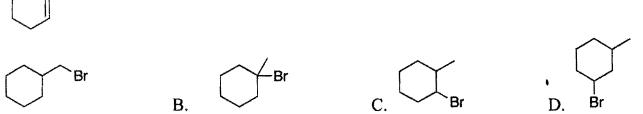
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6. Which of the following alkyl halides would afford the indicated product upon reaction with sodium ethoxide?



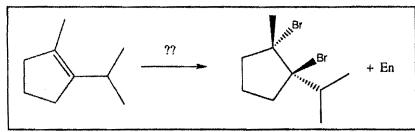
7. Which of the molecules below arises from anti-Markovnikov hydrohalogenation with HBr of the alkene shown?



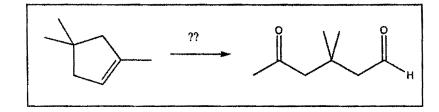
8. Which of the alkenes shown below would produce a chirality center upon hydrohalogenation?

$$A.$$
 $B.$ $C.$ CH_3 $D.$ $C.$

9. Provide the reagent(s) required to complete the following transformation:



- A. HBr
- B. Br₂/hv
- C. Br₂
- D. HBr/ROOR
- 10. Predict the reagent(s) required to complete the following transformation:



- A. 1) OsO₄; 2) NaHSO₃, H₂O
- B. 1) Hg(OAc)₂, H₂O; 2) NaBH₄
- C. 1) RCO_3H ; 2) H_3O^+
- D. 1) O₃; 2) DMS

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II. 申論題 (80 pts)

1. Provide the missing intermediates and reagents in the following synthesis. (20 pts)

C
$$\xrightarrow{\text{Li(s)}}$$
 D $\xrightarrow{1. B}$ $\xrightarrow{2. H_3O^{\oplus}}$

2. Draw the product of each of the following reactions. (40 pts)

(a)
$$\begin{array}{c} 1. \text{ KMnO}_4 \\ \text{NaOH. } \frac{1}{2}. \text{ H}^{\odot} \end{array}$$
 (b) $\begin{array}{c} 1. \text{ KMnO}_4 \\ \text{NaOH. } \frac{1}{2}. \text{ H}^{\odot} \end{array}$? $\begin{array}{c} 1. \text{ C}_3. -78 \, ^{\circ}\text{C} \\ \hline 2. \text{ H}^{\odot} \end{array}$? $\begin{array}{c} 1. \text{ C}_3. -78 \, ^{\circ}\text{C} \\ \hline 2. \text{ Zn, HOAc} \end{array}$?

(d) (e) O
$$\frac{1. O_{3.} -78 °C}{2. H_2O_2}$$
? $\frac{1. OsO_{4.}}{2. HIO_4}$?

3. Draw the complete, detailed mechanism for the following reaction. (20 pts)

$$O$$
 CH_3
 H
 CH_3
 CH_3