淡江大學 107 學年度日間部轉學生招生考試試題

系別: 化學工程與材料工程學系 二年級

科目:普通化學

考試日期:7月26日(星期四) 第1節

本試題共 2 大題,

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第一部分:解釋名詞 (每題5分,共30分)

- 1. Isotope
- 2. Dilution
- 3. Hydration
- 4. Combustion
- 5. Equivalence point
- 6. Electrolyte

第二部分:問答題與計算題(共70分)

- 1. Name the following compounds. (15 pts)

 (a) NaCl (b) H₂O₂ (C) NaH (d) O₃ (e) Na₂CO₃
- 2. Draw the Lewis structure of the following compounds. (15 pts)

 (a) CO (b) SF₆ (c) XeF₄
- 3. Consider the titration of 50 mL of 0.1 M acetic acid (HC₂H₃O₂, $K_a = 1.8 \times 10^{-5}$) by 0.1 M NaOH.
 - (a) Calculate the pH of the initial solution of 0.1 M acetic acid. (10 pts)
 - (b) Calculate the pH of the resulting solution after 25 mL of NaOH have been added. (10 pts)
 - (c) Calculate the pH of the resulting solution after 50 mL of NaOH have been added. (10 pts)
 - (d) Calculate the pH of the resulting solution after 60 mL of NaOH have been added. (10 pts)