淡江大學 105 學年度日間部轉學生招生考試試題

系別: 化學組二年級

科目:普通化學

考試日期:7月21日(星期四)第3節

本試題共 4 大題,

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- 1. (20 pts) For electron and photon, describe their wave and particulate properties.
- 2. (20 pts) Explain the following terms:
 - (a) reaction intermediate
 - (b) rate determining step
 - (c) enantiomers
 - (d) lanthanide contraction
- 3. (30 pts) For each of the following molecules, predict the molecular structure (including bond angles), give the expected hybrid orbitals on the central atom, and predict the overall polarity.
 - (a) OF_2 (b) AsF_5 (c) BF_3
- 4. (30 pts) Answer the following questions for N₂.
 - (a) (10 pts) Draw the molecular orbital (MO) energy-level diagram of the ground state. Label the highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO).
 - (b) Give the electron configuration of the 1st excited state.
 - (c) For (a) and (b), calculate the bond order for each.