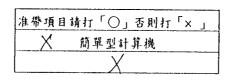
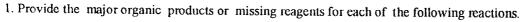
## 淡江大學九十三學年度碩士班招生考試試題

系別:化學學系

科目:有 機 化 學



本試題共 乙



a. 1,3-Butadiene + ? -""<sub>"</sub>соосн<sub>з</sub>

d 
$$O$$
-CH<sub>3</sub>  $O$ -PhCO<sub>3</sub>H  $O$ -P

$$k \bigcirc OCH_3 \longrightarrow OCH_3$$
 1

$$^{\text{m.}}$$
 CH<sub>3</sub>C  $\equiv$ CH  $\xrightarrow{9\text{-BBN}}$   $\xrightarrow{\text{H}_2\text{O}_2, \text{NaOH}}$  ?

n. 
$$CH_3CH=CH-CH=CH_2$$
  $HBr, 60 ^{\circ}C$  ?

o. 
$$\frac{NO_2}{Br_2/FeBr_3}$$
 ?

2. Identify all the C<sub>7</sub>H<sub>16</sub> hydrocarbons that have three peaks in their decoupled C-13 NMR spectrta.

- 6% 3. (A). Which of the following compounds will show IR absorption at 1670~1690 cm<sup>-1</sup>? 3%
  - (B) Which of the following compounds will show no absorption at 3000~3100 cm<sup>-1</sup>? 3%

4. Which one of the following compounds will absorb at longer wavelength in UV region?



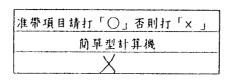
CH<sub>2</sub>OH



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本試題共 2: 頁

5. Which of the following bromides would ionize most rapidly to form a carbocation? Justify your answer.



6. Provide the mechanism for each of the following transformations

a. 
$$Et_2N$$
  $CH_3$   $KOH/H_2O$   $Et_2N$   $CH_3$   $CH_3$ 

b. 
$$\frac{h\nu}{CH_3OH}$$
 COOCH<sub>3</sub>

7. Rank the following compounds in order of reactivity toward esterification with C<sub>2</sub>H<sub>5</sub>OH.

a. CH<sub>3</sub>CH<sub>2</sub>COCl

8. Which one of the following compounds is more polar? Also give your reasons.



a.

9. (a). Which of the following two structures in the following pairs is more stable?

(b). Indicate which is cis fused ring?



10. (a). For the Lewis structure of carbon monoxide shown below, predict the direction of the dipole moment in this Lewis structure.



(b). As it has shown above that carbon monoxide should have a substantial dipole moment. In fact, the experimentally determined dipole moment is very small. 0.11 D. Draw a second resonance structure for carbon monoxide to rationalize the observation.