

# 淡江大學 100 學年度碩士班招生考試試題

27-1

系別：化學學系(化學組)

科目：物理化學

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1)

Based on the physical phenomena of the blackbody radiation we realized that the frequency distribution of the emitted blackbody radiation be described by the function  $R(\nu)$ , where  $R(\nu) d\nu$  is the energy with frequency in the range  $\nu$  to  $\nu + d\nu$  that is radiated per unit time and per unit surface area. In June 1900 Lord Rayleigh attempted to derive the theoretical expression for the  $R(\nu)$  as follow:

$$R(\nu) = (2\pi kT/c^2) \nu^2. \quad \text{Eq. 1-1}$$

On October 1900, the physicist Max Planck announced his derivation of a formula as follows:

$$R(\nu) = (2\pi h/c^2) (\nu^3/e^{h\nu/kt}-1). \quad \text{Eq 1-2}$$

Please make a statement to describe a) both equations Eq. 1-1 and Eq. 1-2 in terms of their different basics (10%), b) how this difference leads to the important concept of the energy quantization. (5%) and c) finally how the value of planck's constant  $h$  is obtained. (5%)

2)

It was realized that the solution of the Schrödinger equation for a particle of mass  $m$  free to move parallel to the X-axis with zero potential energy, that is  $V(x)=0$ , is given as follows:

$$\psi = A e^{ikx} + B e^{-ikx} \text{ and } E = k^2 \hbar^2 / 2m \quad \text{Eq. 2-1}$$

where A and B are constants.

a) Please verify that the  $\psi = A e^{ikx} + B e^{-ikx}$  is indeed a solution of the Schrödinger equation for a particle of mass  $m$  free to move parallel to the X-axis with zero potential energy.(5%)

Having determined the wavefunction  $\psi = A e^{ikx} + B e^{-ikx}$  now we are in a position to calculate the probability density  $|\psi|^2$ .

b) Please evaluate the probability density  $|\psi|^2$  by assuming that either  $B=0$  or  $A=0$ , then make your conclusion about the probability of finding the particle. (5%)

Consider the operators  $d/dx$  and  $d^2/dx^2$ . Is the function  $\psi = A e^{ikx} + B e^{-ikx}$  an eigenfunction of these operators? If so, what are the eigenvalues? Note that A, B and k are real numbers.(5%)

3)

It is realized that for a particle of mass  $m$  freely move anywhere on the surface of a sphere of radius  $r$  is the rotation of  $m$  in three dimensions.

a) If the laplacian  $\nabla^2$  in spherical polar coordinates is written as follows:

$$\nabla^2 = \frac{\partial}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} + \frac{1}{r^2} \Lambda^2 \quad \text{Eq. 3-1}$$

where the legendrian,  $\Lambda^2$  is

$$\Lambda^2 = \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} + \frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \sin \theta \frac{\partial}{\partial \theta} \quad \text{Eq. 3-2}$$

Please write the Schrodinger equation for rotational motion in three dimensions in which the  $r$  is constant. (5%)

b) Please use the separation of variables method to transform the above Schrodinger equation into two parts, that is, the  $\phi$  dependent equation and the  $\theta$  dependent equation, in order to obtain two types of quantum numbers by solving these two equations, respectively. Then please describe the correlation between these two types of quantum numbers. (10%)

4)

Kinetic experiments yield the concentration  $[A]$ .of reacting species as functions of time at a fixed temperature. Suppose the reaction  $aA \rightarrow$  product is the first order with  $r = k [A]$  and the rate law is as follows:

$$r = (-1/a) d[A]/dt = k[A]. \quad \text{Eq. 4-1}$$

a) Please derive expression for the concentration  $[A]$  as a function of time leading to the exact expression of the rate law. (5%)

b) Please rearrange the exact expression for the rate law to illustrate the linear relationship between  $\ln([A]_0/[A])$  versus  $t$  with slope  $ak$ .(5%)

c) Finally, please derive the exact expression of the reaction's half-life  $t_{1/2}$ . (5%)

5)

a) Calculate

$$\left(\frac{\partial f}{\partial x}\right)_y, \left(\frac{\partial f}{\partial y}\right)_x, \left(\frac{\partial^2 f}{\partial x^2}\right)_y, \left(\frac{\partial^2 f}{\partial y^2}\right)_x, \left(\frac{\partial\left(\frac{\partial f}{\partial x}\right)_y}{\partial y}\right)_x \text{ and } \left(\frac{\partial\left(\frac{\partial f}{\partial y}\right)_x}{\partial x}\right)_y$$

For the function  $f(x,y) = ye^x + xy + x \ln y$  (10%)

b) Determine if  $f(x,y)$  is a state function of the variables  $x$  and  $y$ .(5%)

c) if  $f(x,y)$  is a state function of the variables  $x$  and  $y$ , what is the total differential  $df$ ?(5%).

6)

It was realized that the Boltzmann distribution is written as the fraction of molecules in the state  $i$ , that is,  $p_i = n_i / N$ , and it is expressed as  $e^{-\beta \epsilon_i} / q$ .

- a) Please give the name for the  $q$  and write the exact expression for the  $q$ . (5%)
- b) Based on your expression for the  $q$ , please write the  $q$  of the a linear molecule treated as rigid rotor. (5%) **(The energy level of a linear rotor are  $hcBJ(J+1)$  with  $J=0,1,2,3,\dots$  and it consists of  $2J + 1$  degenerate states.)**
- c) Imagine that you have a uniform ladder of energy levels which is similar to the vibrational energy level of a diatomic molecule in the harmonic approximation. c) Please derive the formula for the  $q$ . (5%) **(You probably need this mathematical expression  $1+X+X^2+X^3+X^4+ \dots = 1/(1-X)$ ).**