淡江大學九十三學年度碩士班招生考試試題

系別:化學學系

科目:分析 化學

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- 1. a) As is Figure 1, please describe the detail about the three modes of solid phase microextraction (SPME) operation.
 - b) Describe the applications of solid phase microextraction (SPME).

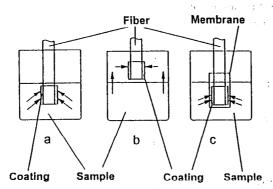


Figure 1 Modes of SPME operation: direct extraction (a), headspace extraction (b) and membrane-protected SPME (c)

2. Calculate the potential of the following cell and indicate whether it is galvanic or electrolytic.

Pt |
$$UO_2^{2+}(0.0150M)$$
, $U^{4+}(0.200M)$, $H^{+}(0.0300M)$ | Fe²⁺(0.0100M), Fe³⁺(0.0250M) | Pt

The two half-reactions are

$$Fe^{3+} + e^{-} \rightleftharpoons Fe^{2+}$$

$$E^{o} = +0.771 \text{ }$$

$$UO_2^{2+} + 4H^+ + 2e^- \rightleftharpoons U^{4+} + 2H_2O$$
 $E^0 = +0.334 \text{ V}$

$$E^0 = +0.334 \text{ V}$$

3. Calculate the molar absorptivity (ϵ_{max}) of $KMnO_4$ using the following data:

A KMnO₄ solution at λ_{max} =522nm gave an absorbance=1.236 in A 10mm cell. The Mn concentration is 30mg/L (Relative atomic masses are

(8%)

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4. Fig. 5.3c shows the UV spectra of azobenzene (Az, concentration 3.73×10^{-3} g dm⁻³) and phenanthrene (P, 3.23×10^{-3} g dm⁻³) both recorded in *iso*-octane on a standard UV/visible spectrophotometer. The wavelength drive on the instrument was 10 nm cm⁻¹ and the absorbance range was 2 a.u.f.s. Measurements were made against *iso*-octane using 10 mm cells.

What wavelength would you choose:

- (i) To detect Az without detecting P;
- (ii) To detect P without detecting Az;
- (iii) To detect both of them;
- (iv) To detect Az at maximum sensitivity?

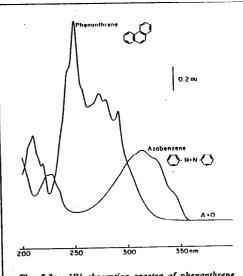


Fig. 5.3c. UV absorption spectra of phenanthrene and azobenzene

(10%)

- 5. How do gas-liquid and gas-solid chromatography differ? (10%)
- 6. Define the following terms (HPLC):
 - a) reversed-phase packing

b) normal-phase packing

(10%)

- c) extra-column broadening
- 7. Suggest a type of liquid chromatography that would be suitable for the separation of

- (b) CH₃CH₂OH and CH₃CH₂CH₂OH,
- (c) Ba^{2+} and Sr^{2+} .

(10%)

- (d) C₄H₉COOH and C₅H₁₁COOH.
- (e), high-molecular-weight glucosides.

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8. What is the principle of micellar electrokinetic capillary chromatography? How does it differ from capillary zone electrophoresis?

(10%)

9. How do the spectra for electron-impact, field ionization, and chemical ionization sources differ from one another (Molecular mass spectrometry)?

(10%)

10. The following data were collected by finding the absorbance (xi) of replicate solutions of a colored complex of iron. Determine the mean and standard deviation for the five data.

(10%)

Solution	xi		
1	0.752	** *	
2	0.756		(N):
3	0.752		$\int_{-\infty}^{N} r^{2} = \frac{\left(\sum_{i=1}^{N} x_{i}\right)}{\left(\sum_{i=1}^{N} x_{i}\right)}$
4	0.751		$s = \sqrt{\frac{\sum_{i=1}^{N} N_i}{\sum_{i=1}^{N} N_i}}$
5	0.760		N-1