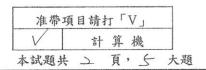
本試題雙面印製

淡江大學 99 學年度碩士班招生考試試題

系別:機械與機電工程學系

科目:熱 力 學



1. As shown in Fig. 1, an oven wall of a 2.5-in-thick layer of steel (κ_s =8.7 Btu/h·ft· $^{\circ}$ R) and a layer of brick (κ_b =0.42 Btu/h·ft· $^{\circ}$ R). At steady state, a temperature decrease of 1.2°F occurs over the steel layer. The inner temperature of the steel layer is 540°F. If the temperature of the outer surface of the brick must be no greater than 105°F, determine the minimum thickness of the brick, in inch, that ensures this limit is met. (20%)

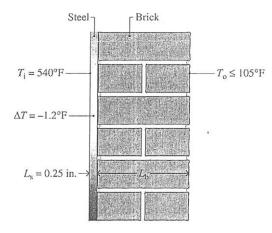


Fig. 1

2. Fig. 2 shows a piston-cylinder assembly fitted with a spring. The cylinder contains water, initially at 1000° F, and the spring is in a vacuum. The piston face, which has an area of 20 in.², is initially at x_1 =20 in. The water is cooled until the piston face is at x_2 =16 in. The force exerted by the spring varies linearly with x according to $F_{Spring} = kx$, where k = 200 lbf/in. Friction between the piston and cylinder is negligible. For the water, determine the work in Btu. (2×10%)

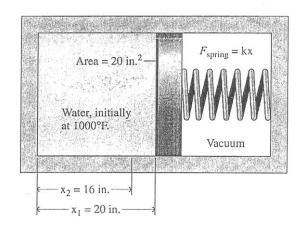
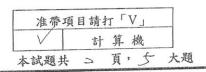


Fig. 2

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3. As shown in Fig. 3, air enters the diffuser of a jet engine operating at steady state at 18 kPa, 216 K and a velocity of 265 m/s, all data corresponding to high-altitude flight. The air flows adiabatically through the diffuser and achieves a temperature of 250 K at the diffuser exit. Using the ideal gas model for air, determine the velocity of the air at the diffuser exit, in m/s.

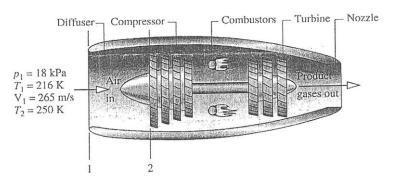


Fig. 3

4. The pressure volume diagram of a Carnot power cycle executed by an ideal gas with constant specific heat ratio k is shown in Fig. 4. Find the relation for V_1 , V_2 , V_3 , V_4 . (20%)

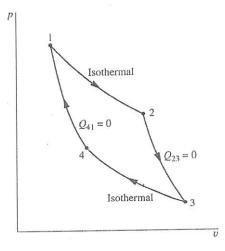


Fig. 4

5. Carbon Dioxide (CO₂) as an ideal gas executes a Carnot cycle while operating between thermal reservoirs at 450 and 100°F. The pressures at the initial and final states of the isothermal expansion are 400 and 200 lbf/in.², respectively. The specific heat ratio is k =1.24. Determine the thermal efficiency.
(20%)