淡江大學 95 學年度碩士班招生考試試題

系別:機械與機電工程學系

科目:熱 力 學

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- 1. Write out the meaning of the following terms: (15%)
 - (a) Thermodynamic system
 - (b) surroundings
 - (c) boundary
 - (d) intensive property
 - (e) extensive property
- 2. Try to explain the following thermodynamic terms: (15%)
 - (a) Kelvin-Planck statement of the second law
 - (b) irreversibilities
 - (c) Kelvin temperature scale
 - (d) Carnot two corollaries of the second law of a power cycle
 - (e) increase of entropy principle
- 3. What are the isentropic efficiencies of turbine, nozzle and compressor? (15%)
- 4. Answer the following true or false. If false, explain why. (15%)
 - (a) A process that violates the second law of thermodynamics violates the first law of thermodynamics.
 - (b) When a net amount of work is done by a closed system undergoing an internally reversible process, a net heat transfer of energy to the system also occurs.
 - (c) One corollary of the second law of thermodynamics states that the change in entropy of a closed system must be greater than zero or equal to zero.
 - (d) A closed system can experience an increase in entropy only when there is energy transfer by heat to the system during the process.
 - (e) Entropy is produced in every internally reversible process of a closed system.

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頁--2 本試題共

- 5. An inventor claims to have developed a power cycle capable of delivering a net work output of 410 kJ for an energy input by heat transfer of 1000 kJ. The system undergoing the cycle receives the heat transfer from hot gases at a temperature of 500 K and discharges energy by heat transfer to the atmosphere at 300 K. Is it possible? (20%)
- 6. What is the working fluid in an ideal Rankine cycle as show. The net power output of the cycle is 100 MW. Determine for the cycle (20%)
 - (a) the mass flow rate of water, in kg/sec.
 - (b) the rate of heat transfer to the working fluid passing through the boiler, in kW.
 - (c) the rate of heat transfer to the working fluid passing through the condenser, in kW.
 - (d) the thermal efficiency.

