

淡江大學九十學年度碩士班招生考試試題

系別：機械工程學系

科目：熱力學 (含熱傳導)

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本試題雙面印製

1. Try to explain the following thermodynamic terms: (15%)
 - (1) internal energy
 - (2) thermal efficiency
 - (3) enthalpy
 - (4) back work ratio
 - (5) specific flow exergy

2. Using temperature-entropy diagram to explain the following thermodynamic cycle: (15%)
 - (1) Air-standard Diesel cycle
 - (2) Ericsson cycle
 - (3) Stirling cycle
 - (4) Air-standard ideal Brayton cycle
 - (5) Air-standard dual cycle

3. Consider as the system an automobile engine. List the irreversibilities present during operation. (15%)

4. Answer the following true or false. If false, explain why. (15%)
 - (a) A process that violates the second law of thermodynamics violates the first law of thermodynamics.
 - (b) When a net amount of work is done by a closed system undergoing an internally reversible process, a net heat transfer of energy to the system also occurs.
 - (c) One corollary of the second law of thermodynamics states that the change in entropy of a closed system must be greater than zero or equal to zero.
 - (d) A closed system can experience an increase in entropy only when there is energy transfer by heat to the system during the process.
 - (e) Entropy is produced in every internally reversible process of a closed system.

◀ 注意背面尚有試題 ▶

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5. A power cycle operating between two reservoirs receives energy Q_H by heat transfer from a hot reservoir at $T_H=2000\text{K}$ and rejects energy Q_C by heat transfer to a cold reservoir at $T_C=400\text{K}$. For each of the following cases determine whether the cycle operates reversibly, irreversibly, or is impossible:

(20%)

- (a) $Q_H=1100\text{kJ}$, $W_{\text{cycle}}=900\text{kJ}$
- (b) $Q_H=1000\text{kJ}$, $Q_C=200\text{kJ}$
- (c) $W_{\text{cycle}}=1400\text{kJ}$, $Q_C=600\text{kJ}$
- (d) $\eta=50\%$

6. During steady-state operation, a parallel-shaft gearbox receives 600kW through the high-speed shaft, but owing to friction and other irreversibilities, delivers 588kW through the low-speed shaft. The gearbox is cooled on its outer surface according to

$$\dot{Q} = -hA(T_b - T_0)$$

where h is the *heat transfer coefficient*, A is the outer surface area, T_b is the uniform temperature of the outer surface, and T_0 is the uniform temperature of the surroundings away from the immediate vicinity of the gearbox. Evaluate the rate of entropy production $\dot{\sigma}$, in kW/K , for (a) the gearbox as the system and (b) an enlarged system consisting of the gearbox and enough of its surroundings that heat transfer occurs at temperature T_0 . Let $h=0.17\text{ kW/m}^2\text{K}$, $A=1.8\text{m}^2$, and $T_0=300\text{K}$.

(20%)