

# 淡江大學九十二學年度碩士班招生考試試題

系列：化學工程與材料工程學系

科目：輸送現象與單元操作

准帶項目請打「○」否則打「×」
○ 簡單型計算機

本試題共 2 頁

本試題雙面印製

1. Water having density of  $998 \text{ kg/m}^3$  is flowing in a channel with a rectangular cross section as shown in Fig. 1. The rectangular cross section has width of  $0.15 \text{ m}$  and height of  $0.1 \text{ m}$ . The viscosity of water is  $0.0089 \text{ kg/m}\cdot\text{s}$ . Calculate the Reynolds number ( $Re$ ) in the channel when the water volume flow rate is (a)  $5.4 \text{ m}^3/\text{hr}$  and (b)  $21.6 \text{ m}^3/\text{hr}$ . (20%)

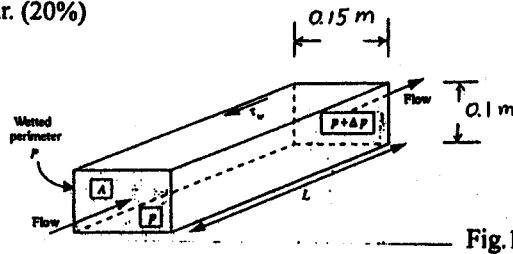


Fig.1

2. Water (density( $\rho$ )= $998 \text{ kg/m}^3$ , viscosity= $0.0089 \text{ kg/m}\cdot\text{s}$ ) is flowing in an inclined pipe as shown in Fig. 2. The pipe length ( $L$ ) is  $50 \text{ m}$  and inner diameter ( $D$ ) is  $0.12 \text{ m}$ , and the inclination angle ( $\theta$ ) is  $30^\circ$ . The wall shear stress ( $\tau_w$ ) is related to the Fanning friction factor ( $f_F$ ) in form of  $\tau_w = f_F \rho u_m^2/2$ , where  $u_m$  is the flow velocity. The roughness of the pipe is  $0.0046 \text{ mm}$ . Calculate the pressure drop ( $p_1 - p_2$ ) when the water flow rate is (a)  $3.6 \text{ m}^3/\text{hr}$  and (b)  $21.6 \text{ m}^3/\text{hr}$ . The friction factor plot is shown as Fig. 3. (20%)

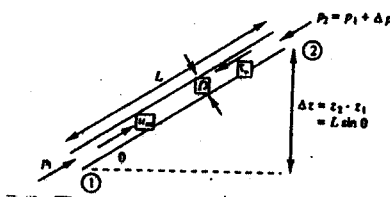


Fig.2

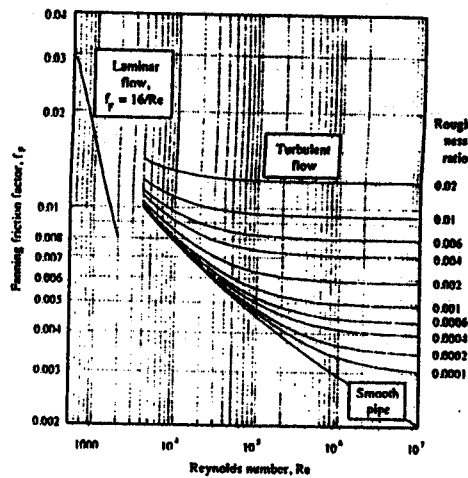


Fig. 3

3. Radial heat transfer is occurring by conduction through a long hollow cylinder of length  $L$  with the ends insulated.

(a) What is the final differential equation for steady-state conduction? The cylinder has a constant thermal conductivity. (5%)

(b) Solve the equation for the temperature profile from part (a) for the boundary conditions given as follows:  $T = T_i$  for  $r = r_i$ ,  $T = T_o$  for  $r = r_o$ . (10%)

(c) Using part (b), derive an expression for the heat flow  $q$ . (5%)

Note: The equation of energy change in cylindrical coordinates is

$$\left( \frac{\partial T}{\partial t} + v_r \frac{\partial T}{\partial r} + \frac{v_\theta}{r} \frac{\partial T}{\partial \theta} + v_z \frac{\partial T}{\partial z} \right) = \alpha \left( \frac{\partial^2 T}{\partial r^2} + \frac{1}{r} \frac{\partial T}{\partial r} + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} \right)$$

# 淡江大學九十二學年度碩士班招生考試試題

系別：化學工程與材料工程學系      科目：輸送現象與單元操作

准帶項目請打「○」否則打「×」
○ 簡單型計算機

本試題共 2 頁

4. A drop of liquid toluene is kept at a uniform temperature of 26°C and is suspended in the air by a fine wire. The initial radius  $r_1 = 2.0$  mm. The vapor pressure of toluene at 26°C is  $p_{A1} = 3.85$  kPa and the total pressure is  $P = 101.325$  kPa. The density of the liquid toluene ( $\rho_A$ ) is 866 kg/m<sup>3</sup> and the molecular weight of toluene ( $M_A$ ) is 92 kg/kgmole. The gas constant  $R$  is 8314 m<sup>3</sup>-Pa/kgmole-K.

(a) Derive the following equation to relate the diffusivity  $D_{AB}$  and the time  $t_F$  for the drop to evaporate completely in a large volume of still air.

$$t_F = \frac{\rho_A r_1^2 R T p_{BM}}{2 M_A D_{AB} P (p_{A1} - p_{A2})}$$

where  $p_{A2}$  is the partial pressure of toluene at a distance  $r_2$  far from the liquid drop, and  $p_{BM} = (p_{A1} - p_{A2}) / \ln[(P - p_{A2}) / (P - p_{A1})]$ . (15%)

(b) Calculate the value of  $D_{AB}$  in unit of m<sup>2</sup>/s, if the time  $t_F$  for complete evaporation is 1400 seconds. (5%)

5. A single-effect evaporator is used for concentrating a feed of 9000 kg/h of a 10 wt% aqueous solution of NaOH to a product of 40 wt% NaOH solution. The feed enters the evaporator at 20°C. The pressure in the vapor space of the evaporator is 11.7 kPa and the pressure of the saturated steam used is 172.2 kPa. The overall heat-transfer coefficient is 1988 W/m<sup>2</sup>-K. Calculate the steam used, the steam economy in kg vaporized/kg steam, and the heating area. (20%)

The boiling point of pure water at 11.7 kPa is 48.9°C, and the enthalpy of saturated vapor at this temperature is 2590 kJ/kg. The heat capacity of the superheated steam is 1.884 kJ/kg-K. For the saturated steam at 172.2 kPa, the temperature is 115.6°C and the latent heat is 2214 kJ/kg. The enthalpy of feed at 20°C is 70 kJ/kg, and the enthalpy of the product (40 wt% NaOH solution) can be calculated using following equation:  $h = 70 + 5.2(T - 20)$ , where  $T$ (°C) is the temperature of the product and  $h$ (kJ/kg) is the enthalpy. The boiling-point rise chart for system NaOH-water is shown in Fig. 4. The reference state of the enthalpy is liquid water at 0°C.

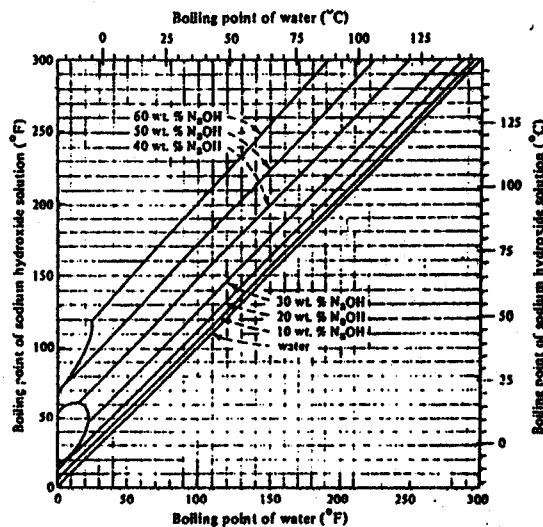


Fig. 4