淡江大學 100 學年度碩士班招生考試試題

58

系別:化學工程與材料工程學系 科目:材 料 科 學

考試日期:2月28日(星期一) 第2節

本試題共 8 大題,共 1 頁

- 1. Please cite the main differences between ionic, covalent and metallic bonding. (10 pts)
- 2. How does the electron structure of an isolated atom differ from that of a solid material? (10 pts)
- 3. Will Cadmium act as a donor or an acceptor when added to a GaAs semiconducting material? Assume that the impurity elements are substitutional. (10 pts)
- 4. Please explain why porosity decreases the thermal conductivity of ceramic and polymeric materials, rendering them more thermally insulative. (10 pts)
- 5. Please explain the two sources of magnetic moments for electrons in an atom. Does each electron have a net magnetic moment? Why or why not?(15 pts)
- 6. Please describe the phenomenon of electronic polarization by electromagnetic radiation. What are two consequences of electronic polarization in transparent materials? (20 pts)
- 7. Please explain why amorphous polymers are transparent, while predominantly crystalline polymers appear opaque or, at best, translucent. (15 pts)
- 8. Please explain why the tendency of a polymer to crystallize decreases with increasing molecular weight. (10 pts)