

淡江大學九十學年度日間部轉學生招生考試試題

系別：機械工程學系三年級

科目：熱 力 學

准帶項目請打「○」否則打「×」	
計算機	字典
○	×

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本試題雙面印製

1. Try to explain the following thermodynamic laws:
- (1) the zeroth law of thermodynamics ,
 - (2) the first law of thermodynamics ,
 - (3) the second law of thermodynamics .
- (15%)

2. Try to explain the following thermodynamic processes:
- (1) reversible process ,
 - (2) irreversible process ,
 - (3) internally reversible process ,
 - (4) isobaric process ,
 - (5) adiabatic process .
- (15%)

3. Using temperature-entropy diagram to explain the following thermodynamic cycles:
- (1) Carnot power cycle ,
 - (2) ideal Rankine cycle ,
 - (3) ideal reheat vapor power cycle ,
 - (4) air standard Otto cycle ,
 - (5) air standard Diesel cycle .
- (15%)

4. Answer the following true or false . If false , explain why .
- (a) A process that violates the second law of thermodynamics violates the first law of thermodynamics .
 - (b) When a net amount of work is done by a closed system undergoing an internally reversible process , a net heat transfer of energy to the system also occurs .
 - (c) One corollary of the second law of thermodynamics states that the change in entropy of a closed system must be greater than zero or equal to zero .
 - (d) A closed system can experience an increase in entropy only when there is energy transfer by heat to the system during the process .
 - (e) Entropy is produced in every internally reversible process of a closed system .
- (15%)

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5. A closed system undergoes a thermodynamic cycle consisting of the following processes :
- Process 1-2 : adiabatic compression with $pV^{1.4} = \text{constant}$ from $p_1 = 50 \text{ lbf/in}^2$, $V_1 = 3 \text{ ft}^3$ to $V_2 = 1 \text{ ft}^3$
 - Process 2-3 : constant volume
 - Process 3-1 : constant pressure, $U_1 - U_3 = 46.7 \text{ Btu}$
- There are no significant changes in kinetic or potential energy .
- (a) sketch the cycle on a p-V diagram .
 - (b) calculate the net work for the cycle , in Btu .
 - (c) calculate the heat transfer for process 2-3 , in Btu .
- (20%)

6. An electric motor operating at steady state draws a current of 10 amp with a voltage of 220V . The output shaft rotates at 1000RPM with a torque of 16 N·m applied to an external load . The rate of heat transfer from the motor to its surroundings is related to the surface temperature T_b and the ambient temperature T_0 by $hA(T_b - T_0)$, where $h = 100 \text{ W/m}^2 \cdot \text{K}$, $A = 0.195 \text{ m}^2$, and $T_0 = 293 \text{ K}$. Energy transfers are considered positive in the directions indicated by the arrows as shown.
- (a) Determine the temperature T_b , in K .
 - (b) For the motor as the system , determine the rate of entropy production , in kW/K.
 - (c) If the system boundary is located to take in enough of the nearby surroundings for heat transfer to take place at temperature T_0 , determine the rate of entropy production , in kW/K , for the enlarged system .
- (20%)

